SYNTHESIS AND CHARACTERISATION OF DEUTERATED CLENBUTEROL AND TWO EQUINE METABOLITES

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ABSTRACT

The synthesis of deuterated clenbuterol (1) and 2 major equine metabolites of clenbuterol 1-(4-amino-3,5-dichlorophenyl)ethane-1,2-diol (2) and 2-[2-(4amino-3,5-dichlorophenyl)-2-hydroxyethylamino]-2methylpropan-1-ol (3) is described. Clenbuterol-do obtained from 1-(4-amino-3,5-W25 (II) dichlorophenyl)-2-bromoethan-1-one (6) after reaction with tert-butylamine-do in tetrahydrofuran (THF) and a subsequent reduction of the obtained aminoketone with sodium borohydride in methanol. Clenbuterol diol metabolite 2 was prepared from 1-(4-amino-3,5-dichlorophenyl)-2-bromoethan-1-one (6) by reaction with sodium acetate followed by reduction of the keto group with sodium borohydride. Analogously, the hydroxy metabolite 3 was obtained after reaction of 1-(4-amino-3,5dichlorophenyl)-2-bromoethan-1-one (6) with 2amino-2-methylpropanol in THF and subsequent reduction of the obtained product with sodium borohydride in methanol. All products were characterised by mass spectrometry and nuclear magnetic resonance.

INTRODUCTION

Clenbuterol is a ß-agonist/antagonist bronchodilator and the only member of this group approved by the US Food and Drug Administration-for use in horses. Because clenbuterol may be used as a therapeutic

medication, in racehorses, particularly for bronchospasm, it is classified by the Association of Racing Commissioners International (ARCI) as a Class 3 agent, and its detection in post performance samples may lead to sanctions against trainers.

To develop a highly sensitive and specific analytical method for clenbuterol (Lehner et al. 2001) the synthesis of deuterated clenbuterol (clenbuterol-d₂) and its 2 major equine metabolites 2 and 3 (J. F. Quirke, Boeringer Ingelheim Vetmedica GmBH, personal communication; Schmid et al. 1990) was required.

MATERIALS AND METHODS

2-[2-(4-Amino-3,5-dicblorophenyl)-2-bydroxyetbylamino]-2-metbyl-d₃-propane-d₆ (clenbuterol-d₄) (1)

The synthesis of deuterated clenbuterol was performed following the procedure described by Keck et al. (1972) and using deuterated tert-butyl-dy-amine (Isotec Inc) instead of tert-butylamine. The product was purified upon column chromatography on silica gel and crystallised from ethyl ether.

Characterisation data: Colourless crystals from ethyl ether, Mp. 182-185°C; ¹H-NMR (300 MHz, CDCl₃: 8 (ppm) 2.91 (dd, 1 H, J 10.5 Hz, J 12.0 Hz, CH₂N), 3.18 (dd, 1H, J 2.4 Hz, J 12.0 Hz, CH₂N), 4.48 (s, 2H, NH₂), 5.30 (dd, 1H, J 2.4 Hz, J 10.5 Hz, CHOH), 7.29 (s, 2 H, 2 × H_{AR}).

Fig 1: Structures of deuterated clembuterol-d₉ and 2 major equine metabolites of clembuterol.

Fig 2: Synthesis of 1-(4-amino-3,5-dichlorophenyl)-2-bromoethan-1-one (6).

Fig 3: Synthesis of clenbuterol-do

1-(4-Amino-3,5-dicbloropbenyl)etbane-1,2-diol (2)

chloroform solution of 1-(4-amino-3,5-٨ dichlorophenyl)-2-bromoethan-1-one (6) was added to a concentrated aqueous sodium acetate tetra-n-butylammonium solution containing bromide as phase transfer catalyst. The mixture was stirred vigorously for 3 days at room temperature. The chloroform phase was then separated, washed with water and dried with MgSO4. After evaporation of the solvent, satisfactorily pure 2-(4amino-3,5-dichlorophenyi)-2-oxoethyl acetate (8) was isolated (yield 88%) which, without any further purification, was reduced with NaBH4 in methanol over a period of 24 h. Methanol was evaporated from the reaction mixture and clenbuterol metabolite 2 was purified upon column chromatography.

Characterisation data: Colourless crystals from acetone-hexane. Mp. 90-93°C; 1 H-NMR (300 MHz, CDCl₃): δ (ppm) 1.99 (broad triplet, 1 H, CH₂OH), 2.50 (d, 1 H, CHOH), 3.55-3.75 (m, 2 H, CH₂OH), 4.46 (bs, 2 H, NH₂), 4.65-4.73 (m, 1 H, CHOH), 7.21 (s, 2 H, 2 × H_{AR}); 1 H-NMR (300 MHz, CDCl₃ with a drop of D₂O): δ (ppm) 3.60 (dd, 1 H, J 8.1 Hz, J 11.4 Hz, CH₂OH), 3.71 (dd, 1H, J 3.6 Hz, J 11.4 Hz, CH₂OH), 4.68 (dd, 1 H, J 3.6 Hz, J 8.1 Hz, CHOH), 7.21 (s, 2 H, 2 × H_{AR}).

2-[2-(4-amino-3,5-dicblorophenyl)-2bydroxyetbylamino]-2-metbylpropan-1-ol (3)

The synthesis was performed by reaction of 1-(4-amino-3,5-dichlorophenyl)-2-bromoethan-1-one
(6) with 2-amino-2-methylpropanol in THF and

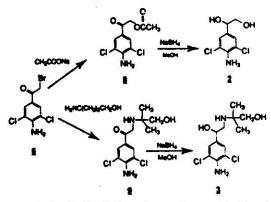


Fig 4: Synthesis of 2 major equine metabolites of clembuterol 2 and 3.

subsequent reduction with NaBH4 following the procedure described by Keck et al. (1972).

Chiracterisation data: Colourless crystals from acetone-hexane Mp. 149-150°C; ¹H-NMR (300 MHz. DMSO-d₆): 8 (ppm) 0.91, 0.92 (2 s, 6 H; 2 x CH₂), 2.57 (m, 2 H, CH₂N), 3.17 (qAB, 2 H, CH₂OH), 4.31 (dd, 1H, J 5.1 Hz, J 7.5 Hz, CHOH), 5.36 (s, 2H, NH₂), 7.19 (s, 2 H, 2 x H_{AR}).

RESULTS AND DISCUSSION

The crucial substrate for the synthesis of all 3 compounds was 1-(4-amino-3,5-dichlorophenyl)-2-bromoethan-1-one (6), which was obtained as shown in Figure 2. Chlorination of the commercially available 4-aminoacetophenone (4µ) in acetic acid gave the 3,5-dichloroderivative 5 (Lutz et al. 1947) which, after bromination in chloroform according to Keck's procedure (Keck et al. 1972), provided bromoketone 6. This compound was the initial substrate for the synthesis of all 3 target compounds: both metabolites of clenbuterol and deuterated clenbuterol-d₀.

Clenbuterol-d₀ was synthesised according to the procedure described by Keck et al. (1972). The reaction of 1-(4-amino-3,5-dichlorophenyl)-2bromoethan-1-one (6) with terr-butyl-d₀-amine in

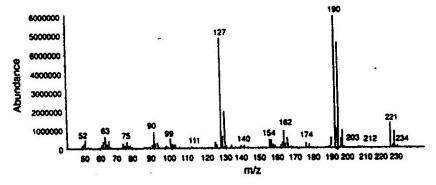


Fig 5: Mass spectrum (El mode) of 1-(4-amino-3,5-dicbloropbenyl)elbane-1,2-diol(2).

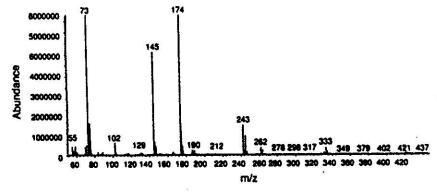


Fig 6: Mass spectrum (El mode) of 2-{2-(4-amino-3,5-dichlorophenyl)-2bydroxyethylaminol-2methylpropan-1-ol (3) derivatised with BSTFA-1%TMCS.

THF afforded 2-{2-(4-amino-3,5-dichlorophenyl)-2-oxoethylamino|-2-methyl-d₃-propane-d₆ (Z) (Fig 3). The reduction of the carbonyl to hydroxy group in Z was performed using sodium borohydride in methanol. Clenbuterol-d₉ (1) was characterised by electron impact MS and ¹H-NMR. The mass spectrum of the TMS derivative of clenbuterol-d₉ revealed the expected isotope shifts showing the molecular ion at m/e 357 of very low abundance and being 9 mass units higher than that of clenbuterol (data not shown).

The glycolic metabolite of clenbuterol was obtained after reaction of 6 (Fig 4) with sodium acetate in a 2 phase solvent system water/chloroform in the presence of phase transfer catalyst tetra-n-butylammonium bromide. The substitution product - acetate 8 was obtained in high yield.

The joint reduction of the ketone and the ester group was made by sodium borohydride in methanol. The ¹H-NMR spectrum of 2 registered in deuterated chloroform is consistent with the assigned structure and significantly simplifies when registering the spectrum after addition of a drop of deuterium water. The glycolic metabolite of clenbuterol was also characterised by mass spectrometry. The electron impact mass spectrum of 2 (Fig 5) contains 2 abundant and characteristic fragment ions at m/z 190 and 127 corresponding to C₇H₆Cl₂NO+ (loss of CH₂OH) and C₆H₆ClN+ respectively. The molecular ion is represented by m/z 221.

The second clenbuterol metabolite 3 was obtained after treating 1-(4-amino-3,5-dichlorophenyl)-2-bromoethan-1-one (6) with 2-amino-2-methylpropanol in THF and reducing the obtained product with sodium borohydride in methanol (Keck et al. 1972, Fig 6).

The El mass spectrum of bis-TMS derivative of 3 contains 3 abundant fragment ions at m/z 145, 174 and 243, where the first 2 correspond to the ions 'Bu-OTMS' and CH₂NH'Bu-OTMS' respectively.

Conclusions

The synthesis of clenbuterol-d₉ and 2 major equine metabolites of clenbuterol was accomplished.

These compounds were designed for analytical methods for clenbuterol. Particularly the deuterium labelled clenbuterol has enabled us to develop a highly pensitive serum test for clenbuterol (Lehner et al. 2001). Use of clenbuterol-d₉ as an internal standard allowed detection of clenbuterol in serum samples at concentrations down to 10 pg/ml.

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